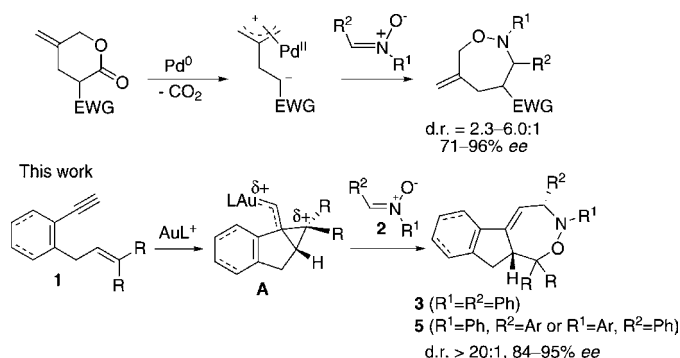


Synthetic Methods

Intermolecular Gold-Catalyzed Diastereo- and Enantioselective [2+2+3] Cycloadditions of 1,6-Enynes with Nitrones**

Sagar Ashok Gawade, Sabyasachi Bhunia, and Rai-Shung Liu*

Metal-mediated cycloaddition reactions are a powerful tool for accessing carbo- and heterocyclic frameworks.^[1] Nitrones serve as three-atom building units in various [3+2] and [3+3] cycloadditions with suitable dipolarophiles to access five- and six-membered nitrogen heterocycles.^[2,3] The utility of such reactions is manifested not only in the easy access to natural products,^[2a] but also in the development of many enantioselective cycloadditions. We are aware of few reports of nitrono cycloadditions for the synthesis of seven-membered heterocycles.^[4] The reported examples are exclusively focused on [4+3] nitrono cycloadditions,^[4–6] as shown by the work of Hayashi and co-workers (Scheme 1).^[4c] We report herein the gold-catalyzed cyclization/[2+2+3] cycloaddition cascade^[5,6] between nitrones and the readily available 1,6-enynes **1** to give the highly substituted 1,2-oxazepane derivatives **3** and **5** with satisfactory diastereo- and enantioselectivity. The importance of this reaction is reflected by the occurrence of 1,2-oxazepane moieties in several bioactive molecules.^[7]

Previous work (Hayashi et al.)^[4c]


Scheme 1. Cycloaddition reactions for nitrones.

For 1,6-enynes, the interception of the 1,4-dipole equivalent **A** is challenging because of its intrinsic reactivity toward cycloisomerization.^[8,9] Echavarren and co-workers attempted to intercept **A** with carbonyl compounds in an intermolecular

fashion,^[10a,c,11] but those reactions were inevitably accompanied by a significant amount of side products.^[10a] Helmchen^[10b] studied the same cycloaddition using unsubstituted 1,6-enynes, but carbonyl compounds other than 2-nitrobenzaldehyde were used in large excess (20 equiv). The use of 1,6-enynes as efficient 1,4-dipole equivalents is only feasible in intramolecular system.^[10c]

We sought to realize the reaction using the 1,6-enyne **1a** (Table 1). The primary task was to reduce the production of 3-

Table 1: Nitrono cycloaddition with various catalysts.

Entry	Catalyst ^[a]	Solvent	3a ^[b] Yield [%]	4 ^[b] Yield [%]
1	AgSbF ₆	DCE	–	46
2	AgNTf ₂	DCE	–	43
3	[PPh ₃ AuCl]/AgSbF ₆	DCE	39	15
4	[PPh ₃ AuCl]/AgNTf ₂	DCE	35	20
5	[LAuCl]/AgSbF ₆	DCE	95	–
6	[LAuCl]/AgNTf ₂	DCE	86	–
7	[IPrAuCl]/AgSbF ₆	DCE	75	15
8	[IPrAuCl]/AgNTf ₂	DCE	72	18
9	[LAuCl]/AgSbF ₆	DCM	87	–
10	[LAuCl]/AgSbF ₆	1,4-dioxane	82	–
11	[LAuCl]/AgSbF ₆	toluene	71	–

[a] L = P(*t*Bu)₂(*o*-biphenyl), [substrate] = 0.19 m. [b] Product yields are reported for compounds isolated after purification on silica gel. DCE = 1,2-dichloroethane, IPr = 1,3-bis(diisopropylphenyl)imidazol-2-ylidene, Tf = trifluoromethanesulfonyl.

isobutenyl-1*H*-indene (**4**), which results from a competitive cycloisomerization.^[12] In a standard operation, the nitrono **2a** (ca. 0.38 m) was first stirred with the catalyst (5 mol %) in DCE (25 °C) for five minutes, and then **1a** in DCE (equal volume) was slowly added to this solution, using a syringe pump, over a 1 hour period.^[13] The conversion was complete at the end of the addition. As shown in entries 1 and 2, we obtained only **4** in 46 and 43 % yield using AgSbF₆ and AgNTf₂, respectively. Cationic gold complexes, [PPh₃AuCl]/AgX (X = SbF₆, NTf₂), gave the nitrono cycloadduct **3a** as a single diastereomer in moderate yield (35–39 %) together with **4** in 15–20 % yield (entries 3 and 4). Pleasingly, the use of [P(*o*-biphenyl)(*t*Bu)₂AuCl]/AgX (X = SbF₆, NTf₂) gave the desired **3a** in 95 and 86 % yield (entries 5 and 6). [IPrAuCl]/AgX (X = SbF₆, NTf₂) gave **3a** in 75–72 % yield in addition to a small amount of **4** (15–18 %; entries 7 and 8). For

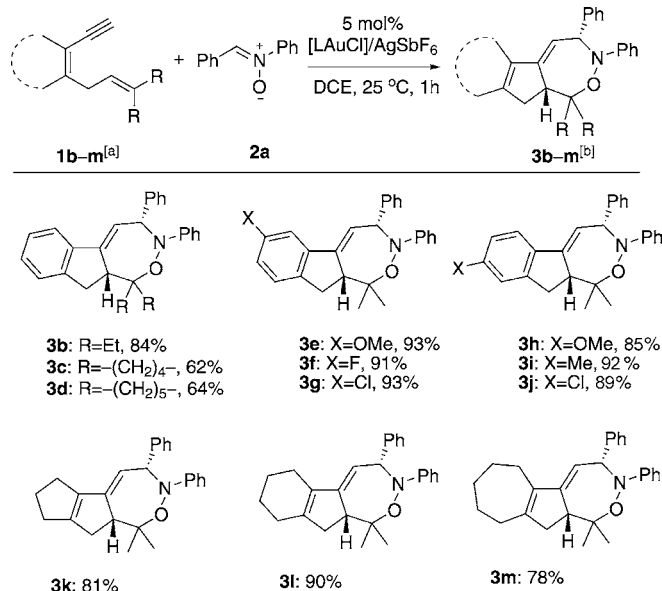
[*] S. A. Gawade, Dr. S. Bhunia, Prof. Dr. R.-S. Liu
Department of Chemistry, National Tsing Hua University
Hsinchu, 30013 (Taiwan)
E-mail: rslu@mx.nthu.edu.tw

[**] We thank the National Science Council, Taiwan, for financial support of this work.

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/ange.201203507>.

[(*o*-biphenyl)P(*t*Bu)₂AuSbF₆], the yields of **3a** depended on solvents: CH₂Cl₂ (87%), 1,4-dioxane (82%), and toluene (71%). Structural characterization of **3a** relied on an X-ray diffraction study^[14] (see the Supporting Information).

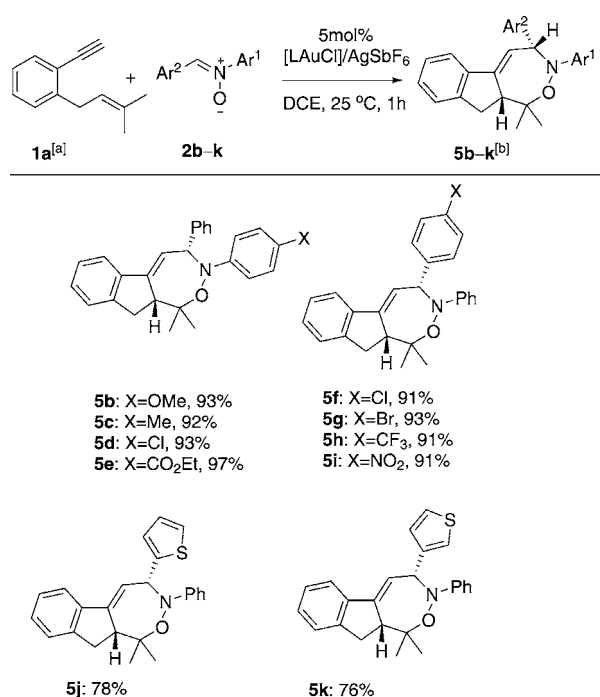
We examined the reactions of **2a** with various 1,6-enynes (**1b–m**) to assess the substrate scope, including benzenoid and nonbenzenoid bridges (Scheme 2). The cycloaddition reac-



Scheme 2. [2+2+3] Nitron cycloadditions on various 1,6-enynes. [a] [substrate]=0.19 M. [b] All product yields are reported for compounds isolated after purification on silica gel. L=P(*t*Bu)₂(*o*-biphenyl), DCE=1,2-dichloroethane.

tions were performed in DCE (25 °C) using [P(*o*-biphenyl)-(*t*Bu)₂AuCl]/AgSbF₆ (5 mol %). This cycloaddition is applicable to substrates bearing various alkenes, such as 3-pentylidene, cyclopentylidene, and cyclohexylidene, thus giving the desired cycloadducts **3b–d** in 62–84% yield. For the benzenoid substrates **1e–g** bearing a C4 substituent (X=OMe, F, Cl), the reactions proceeded smoothly to give the desired compounds **3e–g** with yields exceeding 91%. Excellent yields (85–92%) were obtained also for products **3h–j** bearing various C5 substituents (X=OMe, Me, Cl). The scope of this cycloaddition is substantially expanded with its applicability to nonbenzenoid substrates containing a cyclopentenyl, cyclohexenyl, and cycloheptenyl bridge, thus giving the corresponding products **3k–m** in 78–90% yield.

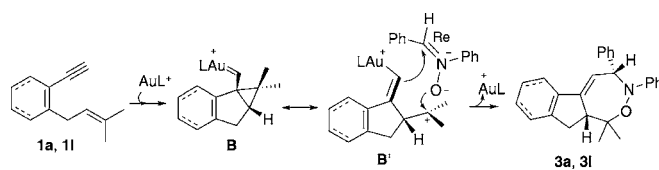
Scheme 3 shows the applicability of the cycloaddition of the 1,6-enyne **1a** with various nitrones (**2b–k**). We obtained only the desired cycloadducts **5b–k** as single diastereomers with yields exceeding 76%. We tested the cycloadditions on nitrones **2b–e**, bearing both electron-deficient and electron-rich aniline moieties (Ar¹=4-XC₆H₄; X=OMe, Me, Cl, CO₂Et), and the corresponding cycloadducts **5b–e** were obtained in excellent yields (92–97%). These cycloadditions can also be extended to the nitrones **2f–i** bearing electron-deficient imines (Ar²=4-XC₆H₄; X=Cl, Br, CF₃, NO₂), thus giving the corresponding cycloadducts **5f–i** with yields of



Scheme 3. Cycloadditions of enyne **1a** with various nitrones. [a] [substrate]=0.19 M. [b] All product yields are reported for compounds isolated after purification on silica gel. L=P(*t*Bu)₂(*o*-biphenyl).

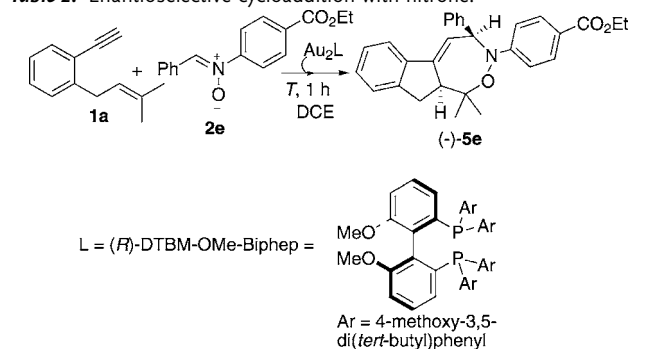
greater than 91%. For the nitrones **2j** and **2k** bearing a 2- and 3-thienyl group, respectively, the corresponding cycloadducts **5j** and **5k** were obtained in 76–78% yield.

Scheme 4 shows a plausible mechanism to rationalize the stereochemistry of the resulting cycloadducts **3a** and **3l** using the 1,6-enynes **1a** and **1l**, respectively, and nitron **2a**. The substrates **1a** and **1l** are less flexible in conformation and facilitate the initial cyclization. In the presence of a gold catalyst, we postulate that an initially formed cyclopropylgold carbenoid^[4–5] has a strong alkenylgold carbocation character (**B'**),^[15] which is attacked by nitron **2a** in a concerted pathway, thus giving either the cycloadduct **3a** or **3l** with excellent diastereoselectivity.



Scheme 4. A proposed mechanism for the diastereoselective [2+2+3] cycloaddition.

Our results for the enantioselective cycloaddition of the 1,6-enyne **1a** with nitron **2e** are shown in Table 2. Among the commonly used C2-bisphosphine ligands, (*R*)-DTBM-MeO-Biphep gave the best performance for enantioselectivity (see the Supporting Information). The use of [LAu₂Cl₂]/2AgSbF₆ (L=(*R*)-DTBM-MeO-Biphep, 5 mol %) gave the desired compound (–)-**5e** in 71% yield with 70% *ee*

Table 2: Enantioselective cycloaddition with nitron.


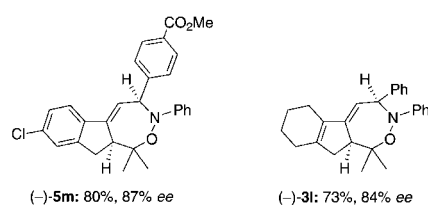
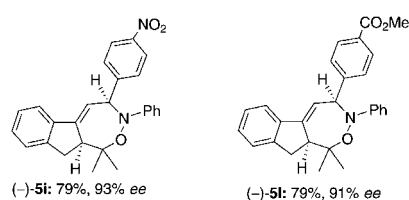
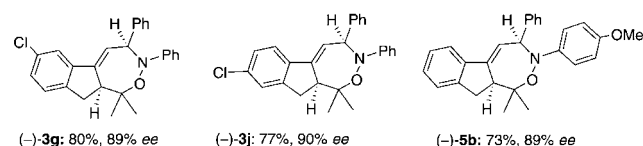
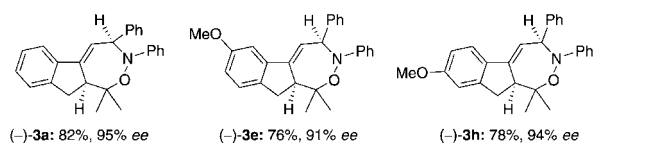
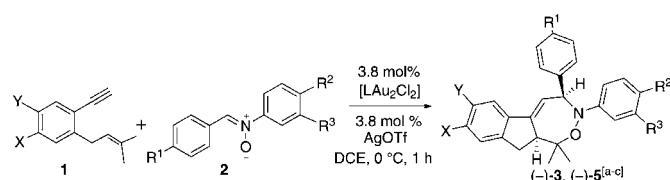
Entry	Catalyst (mol %) ^[a,b]	T [°C]	9e Yield [%]	ee [%] ^[c]
1	[LAu ₂ Cl ₂] (5)/AgSbF ₆ (10)	25	71	(–)-70
2	[LAu ₂ Cl ₂] (5)/AgOTf (10)	25	73	(–)-75
3	[LAu ₂ Cl ₂] (5)/AgOTf (10)	0	71	(–)-76
4	[LAu ₂ Cl ₂] (5)/AgOTf (5)	0	76	(–)-84
5	[LAu ₂ Cl ₂] (3.8)/AgOTf (3.8)	0	74	(–)-92

[a] [substrate] = 0.19 m. [b] All product yields are reported for compounds isolated after purification on silica gel. [c] The ee values were determined by HPLC using a Chiralpak OD-H column.

(entry 1). [LAu₂Cl₂]/2 AgOTf at a 5 mol % loading improved the enantioselectivity of compound (–)-5e to 75% ee (entry 2). The reaction in a cold DCE solution (0°C) had little influence on its ee value (entry 3). However, a change of the Ag/Au ratio to 1:1 for [LAu₂Cl₂]/AgOTf increased the ee value to 84% at 0°C; here, one Au–Cl bond in gold catalyst remains intact (entry 4). We managed eventually to obtain (–)-5e with 92% ee and 74% yield at a low loading catalyst loading (Au/Ag = 1:1, 3.8 mol %, entry 5). We assigned the absolute configuration of (–)-5e according to an X-ray diffraction study on its analogue (–)-5m (see the Supporting Information).

With [LAu₂Cl₂]/AgOTf (3.8 mol %, L = (R)-DTBM-MeO-Biphep), we conducted enantioselective cycloadditions on additional 1,6-enynes and nitrones (Scheme 5). The initial results show the enantioselectivity of the prototype product (–)-3a for which the ee value was up to 95%. We tested this chiral gold catalyst on the benzenoid substrates 1e, 1g, 1h, and 1j, and their corresponding products (–)-3e, (–)-3g, (–)-3h, and (–)-3j were obtained with high ee values (89–94%). We studied the reaction of nitrones bearing an electron-rich aniline moiety and the resulting product (–)-5b was obtained in 73% yield and 89% ee. The same reactions on several nitrones bearing an electron-deficient imine produced the cycloadducts (–)-5i, (–)-5l, and (–)-5m with satisfactory ee values (87–93%). For the nonbenzenoid substrate 1l, the resulting cycloadduct (–)-3l was obtained in 84% ee. X-ray diffraction of compound (–)-5m was performed to clarify its absolute configuration.

Before this work, there was no report on the [2+2+3] cycloadditions of nitrones. The use of a 1,6-enyne as a 1,4-dipole equivalent remains a challenging task. We have developed a new diastereoselective cyclization/[2+2+3] cycloaddition cascade between 1,6-enynes and nitrones. The reactions are applicable not only to a broad range of nitrones,



Scheme 5. Gold-catalyzed enantioselective cycloadditions. [a] [substrate] = 0.19 m. [b] All product yields are reported for compounds isolated after purification on silica gel. [c] The ee values were determined by HPLC using a Chiralpak OD-H or AD-H column. L = (R)-DTBM-MeO-Biphep.

but also 1,6-enynes of benzenoid and nonbenzenoid systems. We also report the success on the enantioselective synthesis of cycloadducts using [LAu₂Cl₂]/AgOTf (L = (R)-DTBM-MeO-Biphep). Most nitron cycloadducts were obtained with high enantiopurity.

Received: May 7, 2012

Published online: June 25, 2012

Keywords: cycloaddition · fused-ring systems · gold · heterocycles · synthetic methods

- [1] For metal-catalyzed cycloaddition reactions, see the selected reviews: a) P. A. Wender, V. A. Verma, T. H. Paxton, T. H. Pillow, *Acc. Chem. Res.* **2008**, *41*, 40–49; b) M. Lautens, W. Klute, W. Tam, *Chem. Rev.* **1996**, *96*, 49–92; c) P. A. Inglesby, P. A. Evans, *Chem. Soc. Rev.* **2010**, *39*, 2791–2805.
- [2] For [3+2] cycloadditions of nitrones, see the selected reviews: a) *Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry Toward Heterocycles and Natural Products* (Eds.: A. Padwa, W. H. Pearson), Wiley, New York, **2002**; b) L. M. Stanley, M. P.

- Sibi, *Chem. Rev.* **2008**, *108*, 2887–2902; c) K. V. Gothelf, K. A. Jørgensen, *Chem. Rev.* **1998**, *98*, 863–910; d) F. Cardona, A. Goti, *Angew. Chem.* **2005**, *117*, 8042–8045; *Angew. Chem. Int. Ed.* **2005**, *44*, 7832–7835.
- [3] For [3+3] cycloadditions of nitrones, see the selected examples: a) I. S. Young, M. A. Kerr, *Angew. Chem.* **2003**, *115*, 3131–3134; *Angew. Chem. Int. Ed.* **2003**, *42*, 3023–3026; b) C. A. Carson, M. A. Kerr, *Angew. Chem.* **2006**, *118*, 6710–6713; *Angew. Chem. Int. Ed.* **2006**, *45*, 6560–6563; c) I. S. Young, M. A. Kerr, *J. Am. Chem. Soc.* **2007**, *129*, 1465–1469; d) F. Liu, D. Qian, L. Li, X. Zhao, J. Zhang, *Angew. Chem.* **2010**, *122*, 6819–6822; *Angew. Chem. Int. Ed.* **2010**, *49*, 6669–6672; e) F. Liu, Y. Yu, J. Zhang, *Angew. Chem.* **2009**, *121*, 5613–5616; *Angew. Chem. Int. Ed.* **2009**, *48*, 5505–5508; f) R. Shintani, S. Park, W.-L. Duan, T. Hayashi, *Angew. Chem.* **2007**, *119*, 6005–6007; *Angew. Chem. Int. Ed.* **2007**, *46*, 5901–5903; g) R. Shintani, T. Hayashi, *J. Am. Chem. Soc.* **2006**, *128*, 6330–6331.
- [4] a) J. Baran, H. Mayr, *J. Am. Chem. Soc.* **1987**, *109*, 6519–6521; b) A. C. Stevens, C. Palmer, B. L. Pagenkopf, *Org. Lett.* **2011**, *13*, 1528–1531; c) R. Shintani, M. Murakami, T. Hayashi, *J. Am. Chem. Soc.* **2007**, *129*, 12356–12357.
- [5] For reviews on gold-catalyzed cycloaddition reactions, see: a) A. S. K. Hashmi, *Chem. Rev.* **2007**, *107*, 3180–3211; b) A. Fürstner, *Chem. Soc. Rev.* **2009**, *38*, 3208–3221; c) N. T. Patil, Y. Yamamoto, *Chem. Rev.* **2008**, *108*, 3395–3442; d) S. M. Abu Sohel, R.-S. Liu, *Chem. Soc. Rev.* **2009**, *38*, 2269–2281; e) F. López, J. L. Mascareñas, *Beilstein, J. Org. Chem.* **2011**, *76*, 1075–1094; f) C. Aubert, L. Fensterbank, P. Garcia, M. Malacria, A. Simonneau, *Chem. Rev.* **2011**, *111*, 1954–1993.
- [6] For gold-catalyzed [4+3] cycloaddition reactions, see: a) I. Alonso, H. Faustino, F. López, J. L. Mascareñas, *Angew. Chem.* **2011**, *123*, 11698–11702; *Angew. Chem. Int. Ed.* **2011**, *50*, 11496–11500; b) T. Wang, J. Zhang, *Chem. Eur. J.* **2011**, *17*, 86–90; c) Y. Bai, J. Fang, J. Ren, Z. Wang, *Chem. Eur. J.* **2009**, *15*, 8975–8978; d) Y. Zhang, F. Liu, L. Zhang, *Chem. Eur. J.* **2010**, *16*, 6146–6150.
- [7] a) T. Kurihara, Y. Sakamoto, H. Matsumoto, N. Kawabata, S. Harusawa, R. Yoneda, *Chem. Pharm. Bull.* **1994**, *42*, 475–480; b) J. I. Levin, A. M. Venkatesan, P. S. Chan, T. K. Bailey, G. Vice, J. Coupet, *Bioorg. Med. Chem. Lett.* **1994**, *4*, 1819–1824; c) I. Takano, I. Yasuda, M. Nishijima, Y. Hitotsuyanagi, K. Tanaka, H. Itokawa, *J. Org. Chem.* **1997**, *62*, 8251–8254.
- [8] Selected reviews: a) E. Jiménez-Núñez, A. M. Echavarren, *Chem. Rev.* **2008**, *108*, 3326–3350; b) D. J. Gorin, B. D. Sherry, F. D. Toste, *Chem. Rev.* **2008**, *108*, 3351–3378; c) A. S. K. Hashmi, M. Rudolph, *Chem. Soc. Rev.* **2008**, *37*, 1766–1775; d) A. Fürstner, P. W. Davies, *Angew. Chem.* **2007**, *119*, 3478–3519; *Angew. Chem. Int. Ed.* **2007**, *46*, 3410–3449; e) L. Zhang, J. Sun, S. A. Kozmin, *Adv. Synth. Catal.* **2006**, *348*, 2271–2296.
- [9] Selected examples: a) E. Jiménez-Núñez, K. Molawi, A. M. Echavarren, *Chem. Commun.* **2009**, 7327–7329; b) A. Fürstner, P. Hannen, *Chem. Eur. J.* **2006**, *12*, 3006–3019; c) X. Linghu, J. J. Kennedy-Smith, F. D. Toste, *Angew. Chem.* **2007**, *119*, 7815–7817; *Angew. Chem. Int. Ed.* **2007**, *46*, 7671–7673; d) A. S. K. Hashmi, L. Ding, J. W. Bats, P. Fischer, W. Frey, *Chem. Eur. J.* **2003**, *9*, 4339–4345; e) L. Zhang, S. A. Kozmin, *J. Am. Chem. Soc.* **2005**, *127*, 6962–6963; f) J.-M. Tang, S. Bhunia, S. M. A. Sohel, M.-Y. Lin, H.-Y. Liao, S. Datta, A. Das, R.-S. Liu, *J. Am. Chem. Soc.* **2007**, *129*, 15677–15683.
- [10] a) A. Escribano-Cuesta, V. Lopez-Carrillo, D. Janssen, A. M. Echavarren, *Chem. Eur. J.* **2009**, *15*, 5646–5650; b) M. Schelwies, A. L. Dempwolff, F. Rominger, G. Helmchen, *Angew. Chem.* **2007**, *119*, 5694–5697; *Angew. Chem. Int. Ed.* **2007**, *46*, 5598–5601; c) E. Jiménez-Núñez, C. K. Claverie, C. Nieto-Oberhuber, A. M. Echavarren, *Angew. Chem.* **2006**, *118*, 5578–5581; *Angew. Chem. Int. Ed.* **2006**, *45*, 5452–5455.
- [11] The cycloadditions of 1,5-enynes with carbonyl compounds do not demonstrate good reaction chemoselectivity; see Ref. [10a].
- [12] R. J. Madhushaw, C.-Y. Lo, C.-W. Hwang, M.-D. Su, H.-C. Shen, S. Pal, I. R. Shaikh, R.-S. Liu, *J. Am. Chem. Soc.* **2004**, *126*, 15560–15565.
- [13] This addition method is crucial for the desired [4+3] cycloaddition. We obtained the cycloisomerization product **4** predominantly (ca. 85%) when a mixture of **1a** and **2a** was stirred together with [P(*o*-biphenyl)(*t*Bu)₂AuCl]/AgSbF₆ in 1,2-dichloroethane.
- [14] The X-ray crystallographic data are provided in the Supporting Information.
- [15] a) A. Fürstner, L. Morency, *Angew. Chem.* **2008**, *120*, 5108–5111; *Angew. Chem. Int. Ed.* **2008**, *47*, 5030–5033; b) D. Benitez, N. D. Shapiro, E. Tkathouk, Y. Wang, W. A. Goddard III, F. D. Toste, *Nat. Chem.* **2009**, *1*, 482–486; c) G. Seidel, R. Mynott, A. Fürstner, *Angew. Chem.* **2009**, *121*, 2548–2551; *Angew. Chem. Int. Ed.* **2009**, *48*, 2510–2513; d) C. Nieto-Oberhuber, M. P. Muñoz, E. Buñuel, C. Nevado, D. J. Cárdenas, A. M. Echavarren, *Angew. Chem.* **2004**, *116*, 2456–2460; *Angew. Chem. Int. Ed.* **2004**, *43*, 2402–2406.